

Tunnelling Characteristics of Stone-Wales Defects in Monolayers of Sn and Group-V Elements

This content has been downloaded from IOPscience. Please scroll down to see the full text.

Download details:

IP Address: 14.139.242.3

This content was downloaded on 31/07/2017 at 09:43

Manuscript version: Accepted Manuscript

Jamdagni et al

To cite this article before publication: Jamdagni et al, 2017, J. Phys.: Condens. Matter, at press:

<https://doi.org/10.1088/1361-648X/aa7dd1>

This Accepted Manuscript is: © 2017 IOP Publishing Ltd

During the embargo period (the 12 month period from the publication of the Version of Record of this article), the Accepted Manuscript is fully protected by copyright and cannot be reused or reposted elsewhere.

As the Version of Record of this article is going to be / has been published on a subscription basis, this Accepted Manuscript is available for reuse under a CC BY-NC-ND 3.0 licence after the 12 month embargo period.

After the embargo period, everyone is permitted to copy and redistribute this article for non-commercial purposes only, provided that they adhere to all the terms of the licence

<https://creativecommons.org/licences/by-nc-nd/3.0>

Although reasonable endeavours have been taken to obtain all necessary permissions from third parties to include their copyrighted content within this article, their full citation and copyright line may not be present in this Accepted Manuscript version. Before using any content from this article, please refer to the Version of Record on IOPscience once published for full citation and copyright details, as permission will likely be required. All third party content is fully copyright protected, unless specifically stated otherwise in the figure caption in the Version of Record.

When available, you can view the Version of Record for this article at:

<http://iopscience.iop.org/article/10.1088/1361-648X/aa7dd1>

1
2
3
4
5
6
7 **Tunnelling Characteristics of Stone-Wales Defects in**
8
9
10
11 **Monolayers of Sn and Group-V Elements**
12
13
14
15
16

17 Pooja Jamdagni ^{1*}, Ashok Kumar², Anil Thakur³, Ravindra Pandey⁴ and P.K.Ahluwalia¹
18
19

20 ¹*Department of Physics, Himachal Pradesh University, Shimla, H.P. India, 171005*
21
22

23 ²*Centre for Physical Sciences, School of Basic and Applied Sciences, Central University of*
24 *Punjab, Bathinda, India, 151001*
25
26

27 ³*Department of Physics, Govt. P. G. College, Solan, H.P. India, 173212*
28
29

30 ⁴*Department of Physics, Michigan Technological University, Houghton, MI, 49931, USA,*
31 *49931*
32
33
34
35
36
37
38
39

40 (June 22, 2017)
41
42
43
44
45

46 *Corresponding author:
47
48

49 Pooja Jamdagni (j.poojaa1228@gmail.com)
50
51
52
53
54
55
56
57
58
59
60

Abstract

Topological defects in ultrathin layers are often formed during synthesis and processing, thereby, strongly influencing the electronic properties of the layered systems. For the monolayers of Sn and group-V elements, we report the results based on density functional theory determining the role of Stone-Wales (SW) defects in modifying their electronic properties. The calculated results find the electronic properties of Sn monolayer to be strongly dependent on the concentration of SW-defects e.g., defective stanene has nearly zero band gap (≈ 0.03 eV) for the defect concentration of $2.2 \times 10^{13} \text{ cm}^{-2}$ which opens up to 0.2 eV for the defect concentration of $3.7 \times 10^{13} \text{ cm}^{-2}$. In contrast, SW-defects appear to induce conduction states in the semiconducting monolayers of group-V elements. These conduction states act as channels for electron tunnelling, and the calculated tunnelling characteristics show the highest differential conductance for the negative bias with the asymmetric current-voltage characteristics. On the other hand, the highest differential conductance was found for the positive bias in stanene. Simulated STM topographical images of stanene and group-V monolayers show distinctly different features in terms of their cross-sectional views and distance-height profiles. These distinctive features can serve as fingerprints to identify the topological defects in experiments for the monolayers of group-IV and group-V elements.

1. Introduction

Atomically thin two-dimensional (2D) materials have received a great deal of attention in recent years due to their outstanding properties and potential applications [1-3]. For example, multilayer stanene (i.e. Sn monolayer) that has been synthesized in a hexagonal lattice [4-5], exhibit topological insulator properties [6-8]. Similarly, group-V monolayers have gained particular attention [9-16]; e.g., arsenene, an atomic layer of As atoms, exhibits four different phases that switch into a topological insulator by an external electric field [17-18]; multilayer antimonene (i.e. Sb monolayer) shows a higher stability in β -phase [19] among the theoretically predicted phases [20-21]; Bismuthene, Bi monolayer, can be fabricated on a graphene substrate [22].

2D materials are found to possess the so-called topological defects which appear due to regrouping of interatomic bonds without forming vacancies and /or addition of foreign impurity atoms in the lattice, the simplest of such defects is Stone-Wales (SW) defect [23-27]. These topological defects have been observed in the atomic monolayers including graphene [28-32]. It has also been reported that the energy barrier for the formation of SW-defects in buckled configurations is lower than in graphene-like planar configurations [33-34]. Moreover, SW-defects are found to be preferential adsorption centres for chemical functionalization [35].

In order to examine such topological defects in 2D materials, transmission electron microscope (TEM) [24] and scanning tunnelling microscope (STM) [36] are the commonly used experimental techniques. These techniques need to be supplemented with theory which provides an insight into origin and evolution of such defects [37-38]. One of the most useful formulations to describe tunnelling phenomenon in STM was proposed by Tersoff and Hamann [39-40] which

1
2
3 uses Bardeen expression to calculate tunnelling current [41]. The wave function associated with
4 the tip is assumed to be spherically symmetric (s-wave character) and tunnelling current can be
5 calculated as convolution of local density of states (LDOS) of the tip and sample. LDOS in
6 vacuum in terms of the LDOS of sample can be described by the Lang's approximation [42]
7 where wave function in vacuum is allowed to decay exponentially.

8
9
10
11
12
13
14
15
16 The Bardeen, Tersoff and Hamann (BTH) model uses first-order perturbation theory to
17 calculate tunnelling current that allows imaging of surface at atomic resolution [39-40]. The
18 tunnelling characteristics using the BTH formalism have been successfully investigated
19 previously for a wide variety of nanomaterials [38, 43-45]. In the present study, the BTH
20 formalism together with density functional theory method have been used to investigate the
21 current-voltage (I-V) characteristics and topography of SW-defects, in stanene and group-V
22 monolayers. It has been found that SW-defects significantly alter the electronic structure and
23 hence, the tunnelling characteristics of considered systems.

2. Computational Details

24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
Electronic structure calculations were performed using density functional theory (DFT) as
implemented in Vienna ab-initio simulation package (VASP) [46]. Generalized gradient
approximation (GGA) within Perdew-Burke-Ernzerhof (PBE) parameterization is used to
describe exchange-correlation functional. The van der Waals (vdW) interaction terms have been
incorporated in calculations by adding a semi-empirical potential to the conventional Kohn-
Sham DFT energy by using DFT-D2 method of Grimme [47]. We have also included the spin-
orbit coupling (SOC) effects in our calculations as these were found to be important to describe
electronic structure of group-V systems [12, 15, 21-22]. A cut-off energy of 400 eV for the
plane wave basis set and a Monkhorst-Pack mesh of (7x7x1) for Brillouin zone integration

1
2
3 were employed. A 15 Å vacuum region in a (5x5x1) hexagonal supercell along the z-direction
4 was applied. All the structures are fully relaxed, with residual forces smaller than 0.01 eV/Å on
5 each atom. Calculations were performed with varying SW-defect concentrations in the
6 monolayers represented by the (5x5) and (6x6) periodic supercells of stanene and group-V
7 monolayer systems.
8
9
10
11
12
13

14 3. Results and Discussion

15
16
17
18
19 Group-V elements exhibit variety of stable allotropic forms [15-19], though the graphene-
20 like structure for group-V monolayers was considered in the present study. Table 1 lists the
21 calculated structural properties of the monolayers considered which were predicted to be
22 buckled. Our calculated values of buckling parameter (Δ) are in excellent agreement with the
23 previously reported values [Table 1]. Negative value of cohesive energy (E_{coh}) indicates the
24 stability of a given monolayer [48]. Note that the cohesive energy was obtained as: $\frac{E_T - nE_a}{n}$,
25 where E_T is the total energy of a supercell simulating a monolayer, E_a is the energy of a free
26 atom and n is the total number of atoms in a monolayer.
27
28
29
30
31
32
33
34
35
36
37
38

39 SW-defects in the monolayers of group-IV and group-V elements show distinctly different
40 atomic reconstructions at the defective sites; e.g., in stanene, the dimer connecting the two
41 pentagons through heptagon shows strong out-of-plan displacement of atoms as compared to
42 antimonene (Figure 1). The relative formation energy (E_{form}) of defective monolayers decreases
43 in going from phosphorene (1.60 eV) to bismuthene (0.91 eV) [Table 1]. Note that E_{form} was
44 calculated as difference between the total energy of SW-defective and pristine monolayers. The
45 formation energy of a pristine monolayer is taken to be 0 eV as the reference energy in Table 1.
46
47
48
49
50
51
52
53
54

55 The monolayers which are energetically most stable have less tendency to form SW-defects and
56
57
58
59
60

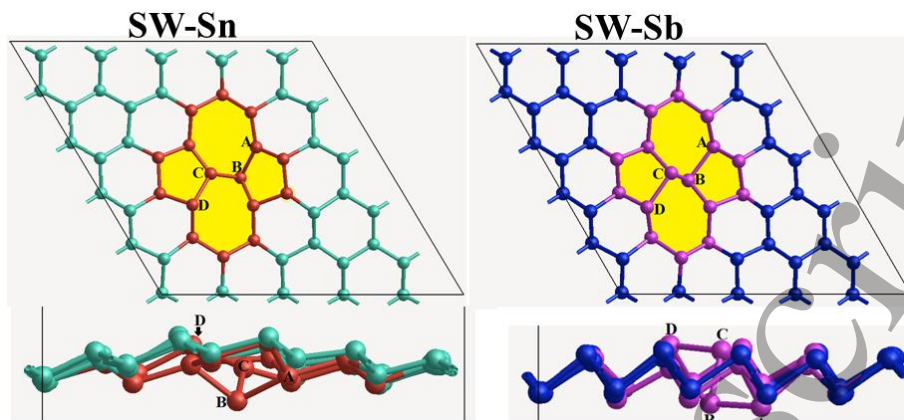


Figure 1: Top and side views of the relaxed structures of stanene (Sn) and antimonene (Sb) with single SW-defect.

Table 1: Buckling parameter (Δ), cohesive energy (E_{coh}), relative formation energy (E_{form}) and band gap (E_{gap}) of pristine and Stone-Wales (SW) defective monolayers in the (5x5) and (6x6) periodic supercells, representing the different defect concentrations in the lattice.

	Pristine monolayers				Defective monolayers							
	Δ (\AA)	E_{coh} (eV)	E_{gap} (eV)		E_{coh} (eV)		E_{form} (eV)		E_{gap} (eV)			
			without SOC	with SOC	(5x5)	(6x6)	(5x5)	(6x6)	without SOC		with SOC	
									(5x5)	(6x6)		(5x5)
Sn	0.85 0.86 ^a	-3.24	0 0.07 0.074 ^a	-3.22 -3.23	1.05 1.27	0.20 0.03	0.20					
P	1.23 1.24 ^b	-5.18	1.97 1.97 ^b	-5.15 -5.16	1.60 1.57	1.53 1.58	1.53					
As	1.39 1.40 ^b 1.39 ^c	-4.49	1.57 1.59 ^b 1.62 ^c	1.41 1.81 ^b	-4.46 -4.47	1.32 1.28	1.25 1.37	1.25			1.23	
Sb	1.64 1.65 ^b	-3.88	1.17 1.26 ^b	0.98 1.00 ^b	-3.85 -3.86	1.06 1.01	1.00 1.05	1.00			0.82	
Bi	1.73 1.71 ^b 1.73 ^d 1.74 ^e	-3.60	0.51 0.55 ^b 0.56 ^d 0.55 ^e	0.50 0.43 ^b 0.50 ^d 0.51 ^e	-3.58 -3.58	0.91 0.94	0.57 0.53	0.57			0.32	

^aRef. [6], ^bRef. [12], ^cRef. [15], ^dRef. [21], ^eRef. [22], ^fRef [34]

1
2
3 vice versa. E_{form} of defective stanene is found to be dependent on the defect concentration; e.g.,
4
5
6 E_{form} changes from 1.27 eV to 1.05 eV as we increase the defect concentration in stanene. On
7
8 the other hand, both relative defect formation energy and cohesive energy of group-V
9
10 monolayers does not depend on SW-defect concentration in the lattice [Table 1].

14 3.1 Electronic Structure

17 Pristine stanene shows a strong spin-orbit coupling (SOC) effect which induces a band gap
18
19 of 0.07 eV by opening the Dirac-like cone at K-point [Figure 2]. Single SW-defect in the (5x5)
20
21 supercell of stanene with the defect concentrations of $2.2 \times 10^{13} \text{ cm}^{-2}$ introduces a band gap of 0.2
22
23 eV. An opening of the gap due to SW-defect is consistent with the previous theoretical studies
24
25 performed on other group-IV monolayers such as silicene and germanene [2, 38]. Both valance
26
27 band maximum (VBM) and conduction band minimum (CBM) are mainly composed of the out-
28
29 of-plan p-orbitals in stanene. On the other hand, an inclusion of the SOC effects in calculations
30
31 find group-V monolayers to be indirect band gap semiconductors [Table 1]. VBM mainly
32
33 consists of the mixture of in-plan and out-of-plan p-orbitals while CBM is dominated by out-of-
34
35 plan p-orbitals [Figure 2]. SW-defects introduce energy levels within the conduction band
36
37 region that result into a reduction of band gap in defective group-V monolayers. The defect
38
39 energy levels mainly consist of p_z -orbitals at the defective site. The valance bands near Fermi
40
41 energy in group-V monolayers show splitting due to the SOC effect. The splitting of bands
42
43 increases in going from arsenene to bismuthene.

44
45
46 The electronic structure of the defective stanene is found to be sensitive to the defect
47
48 concentration e.g., the band gap of the monolayer with SW-defect concentration of 2.2×10^{13}
49
50 cm^{-2} is calculated to be 0.03 eV that changes to 0.2 eV for the defect concentration of 3.7×10^{13}
51
52
53
54
55
56
57
58
59
60

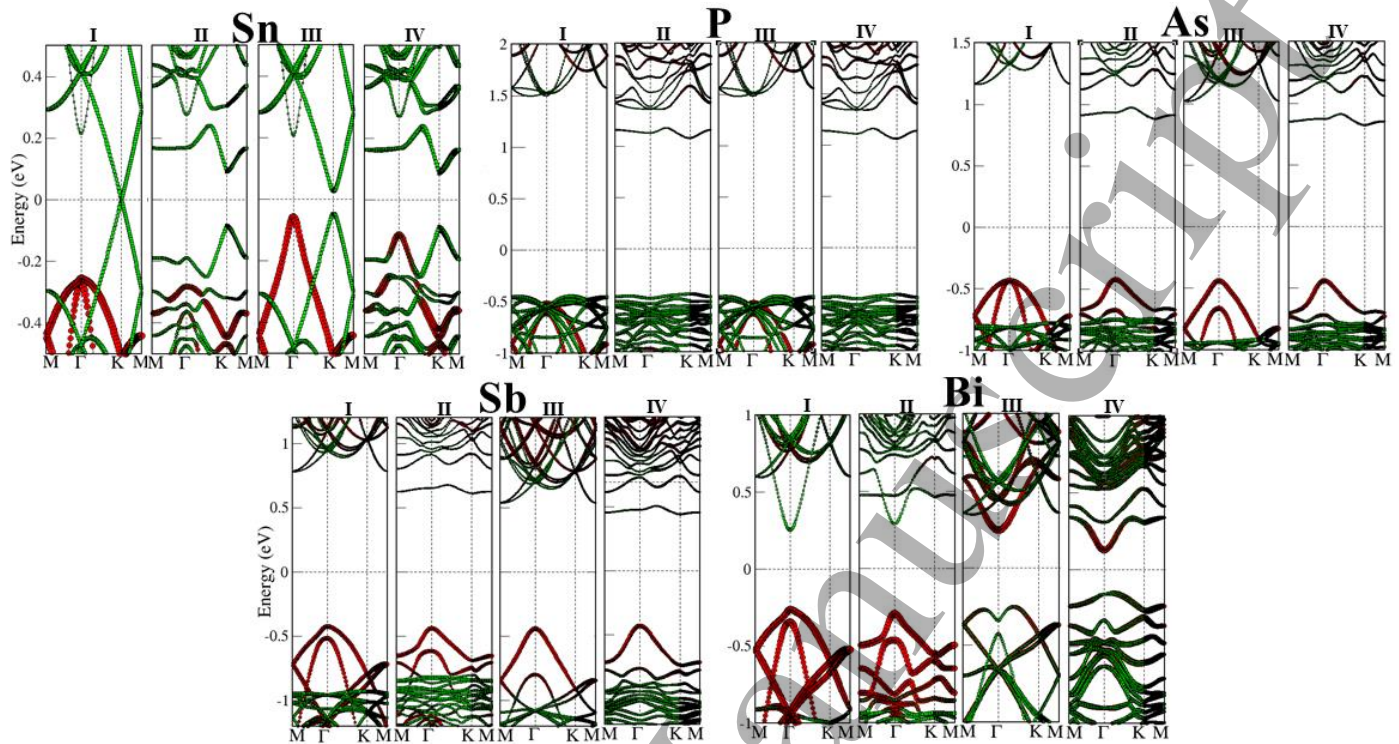


Figure 2: Projected electronic band structure of pristine and SW-defective monolayers of Sn and group-V elements. I and II represent the band structure of pristine and SW-defective monolayers without including spin-orbit coupling (SOC) effect while III and IV represent the corresponding band structure including SOC effect. Red and green colours represent the bands originating from $p_x + p_y$ and p_z orbitals of respective atoms, respectively.

cm^{-2} . On the other hand, the band gap of group-V monolayers remain almost unchanged with the defect concentration in the monolayers [Table 1]; e.g., on increasing the concentration of single SW-defect from $4.4 \times 10^{13} \text{ cm}^{-2}$ to $6.8 \times 10^{13} \text{ cm}^{-2}$ in phosphorene, the band gap value shows very small change from 1.58 eV to 1.53 eV. Note that the hexagonal supercell results in the band edges i.e. VBM and CBM, at K- for the (5x5) supercell case and at Γ -point for the (6x6) supercell case that may result in different band dispersions for temporarily ordered defect configuration [28]. However, the value of the band gap remains the same irrespective of the position of band edges in a given monolayer.

3.2 Tunnelling Characteristics

To study the tunnelling characteristics of the monolayers considered, we have used STM-like setup as shown in [Figure 3\(a\)](#). The Au_{13} cluster is used to model the cap of the tip configuration in the STM setup. We use the BTH formalism to calculate electron tunnelling current [\[49\]](#) which is given by equation (1).

$$I = \frac{4\pi e}{\hbar} \int_{-\infty}^{+\infty} \rho_s\left(\varepsilon + \frac{eV}{2}\right) \rho_t\left(\varepsilon - \frac{eV}{2}\right) e^{-2d\left(2\left(\frac{m}{\hbar^2}\right)(\varphi_{av}-\varepsilon)\right)^{\frac{1}{2}}} \left\{ \left[f\left(\varepsilon - \frac{eV}{2}\right) \right] \left[1 - \left[f\left(\varepsilon + \frac{eV}{2}\right) \right] \right] - \left[f\left(\varepsilon + \frac{eV}{2}\right) \right] \left[1 - \left[f\left(\varepsilon - \frac{eV}{2}\right) \right] \right] \right\} \dots \dots \dots (1)$$

Here, ρ_s and ρ_t are projected density of states (PDOS) of the sample (monolayer) and the tip (Au_{13} cluster), respectively, obtained from DFT calculations. ε is the injection energy of the tunnelling electron and f is the Fermi distribution function. The distance ‘ d ’ is separation between the tip and the sample. The value of the tip-sample separation of 5 Å is taken from the previously reported calculations [\[38, 45, 49-50\]](#). The effective mass of electron (m) and average work function (φ_{av}) of both sample and tip, are assumed to be constant in applied bias (± 1 V). The bias induced changes in the sample’s DOS, which only appear at high enough applied bias [\[49-50\]](#), are not shown in the present study. Note that the BTH formalism is only applicable for small bias voltage (V), $eV \ll \varphi_m$, where φ_m is workfunction of the tip. Considering the average work function of metal, $\varphi_m \sim 4$ eV, the bias voltage limit ± 2 V is typically useful in the BTH formalism [\[51\]](#).

In order to understand the calculated tunnelling characteristics of the monolayers considered, we compared the DOS of sample (monolayer) and tip (Au_{13} cluster) [\[Figure 3\(b\)\]](#).

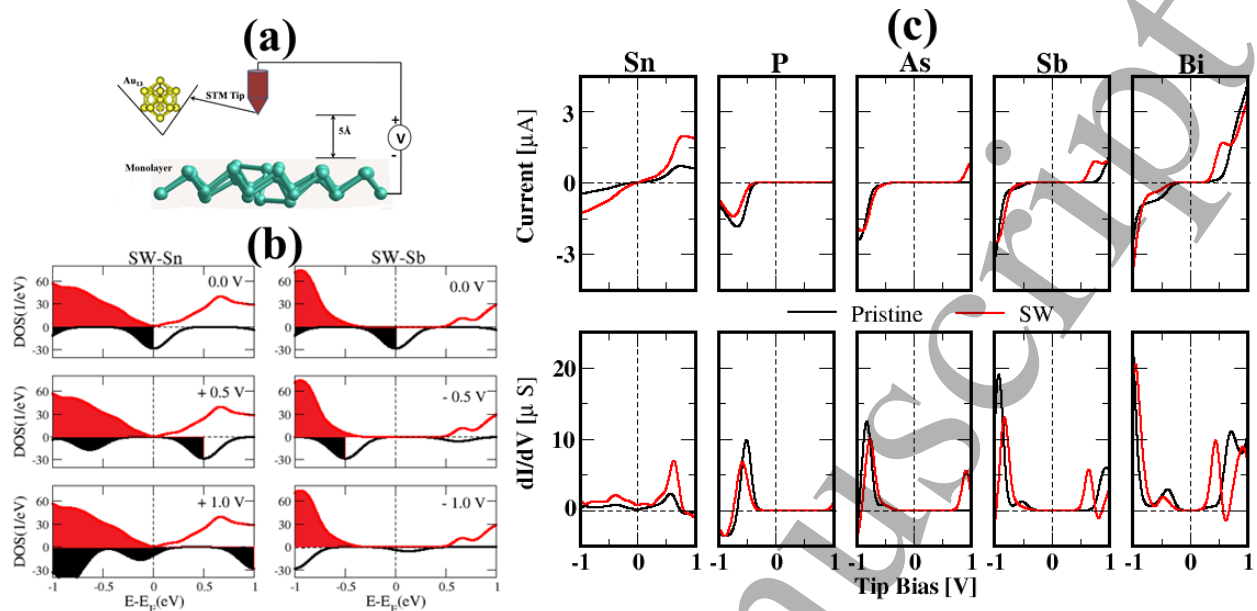


Figure 3: (a) STM-model setup, where distance ‘d’ between tip and sample is taken equal to 5 Å that enters in the exponential part in equation (1). (b) Density of states (DOS) of, stanene and antimonene with SW-defects (in positive y-axis) and Au_{13} tip (in negative y-axis), at different applied bias voltage where black and red shaded regions indicate occupied states of tip and monolayers, respectively. (c) Current-voltage characteristics and differential conductance (dI/dV) of pristine and SW-defective monolayers.

Note that tunnelling current is directly related to the convolution of the local density of states (LDOS) of tip and sample in the relevant regions. For example, none of the electrons of the tip or the sample see open channel to tunnel at zero bias in the defective stanene, so there can be no current. At +0.5 V bias voltage, the electrons within the energies of 0.5 eV above the Fermi level see open channels to tunnel through sample that gives tunnelling current at +0.5 V [Figure 3(c)]. At the bias voltage of +1.0 V, the number of electrons available to tunnel through the sample remains nearly constant that leads to decrease in current [Figure 3(c)]. At very low bias (± 0.3 V), both pristine and defective-stanene show nearly linear current-voltage characteristics which is consistent with their semi-metallic nature.

On the other hand, the pristine group-V monolayers show a semiconducting behaviour. At higher positive bias in the defective group-V monolayers, the magnitude of the tunnelling current increases which is attributed to the available conduction channels facilitated by SW-induced electronic states near the Fermi level. Negative differential conductance (NDC) regions are also found at higher bias [Figure 3(c)] which may be attributed to the less number of available conduction channels to tunnel electrons in these systems. It is underlined that the choice of the tip-sample separation, which is taken to be 5 Å in our calculations, will affect only the magnitude of tunnelling current whereas the features of tunnelling spectra of considered systems remain the same [45, 49].

3.3 Simulated STM Images and Distance-Height Profiles

The STM topographical images can be obtained by applying a small bias voltage (V) between the sample and the tip that produces a tunnelling current whose density, $j(r)$, is the simple extension [52] of the expression derived by Tersoff and Hamann [39-40].

$$j(r, V) \propto \rho_{STM}(r, V) \text{-----(2)}$$

$$\text{where } \rho_{STM}(r, V) = \int_{E_F}^{E_F+eV} dE \rho(r, E) \text{-----(3)}$$

$$\text{and } \rho(r, E) = \sum_{n,k} |\psi_{nk}(r)|^2 \delta(E_{n,k} - E) \text{-----(4)}$$

where $\rho(r, E)$ is local density of states of the tip at r and $\psi_{n,k}$ are the Kohn-Sham eigenstates obtained using density functional theory. Equation (3) describes the tunnelling from the occupied states of a sample to the tip. It is important to emphasize here that for the calculations of STM images, the tip states have been described by s-wave with constant density of states. The tunnelling matrix elements are considered to be independent of the lateral position of the tip for

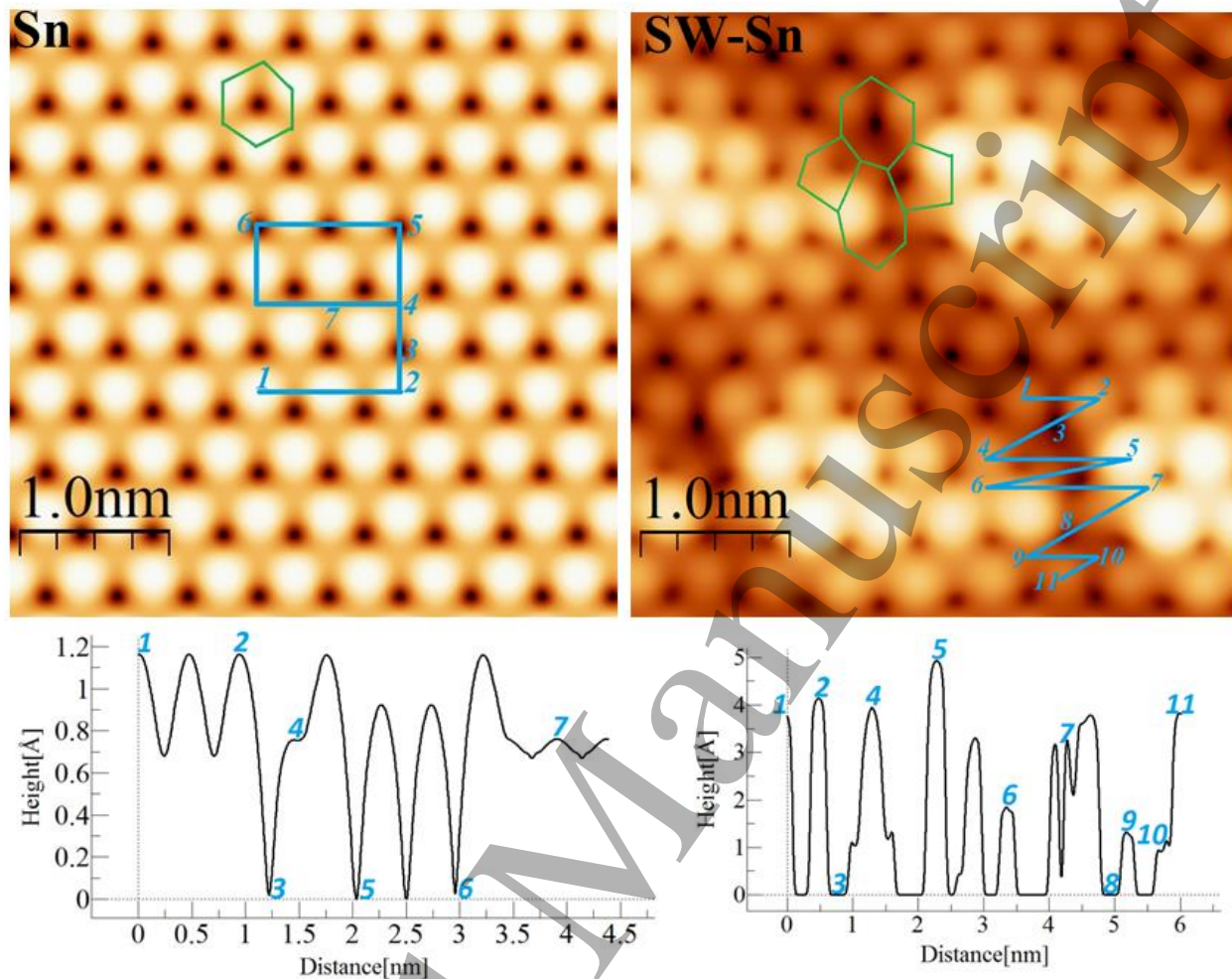


Figure 4: Simulated STM-images of pristine and SW-Sn with distance-height profile along the line shown in STM images, at bias voltage -1 V. Images are obtained at constant height corresponding to current imaging mode with iso-value varies as, $(0 \text{ to } 1.7) \times 10^{-4} \text{ e}/\text{\AA}^3$ in Sn and $(0 \text{ to } 4.8) \times 10^{-4} \text{ e}/\text{\AA}^3$ in SW-Sn. Brightest (darkest) regions in STM images are related with the highest (lowest) peak values in corresponding distance-height curve. The order of iso-value is kept same for all the images for the purpose of comparison.

constant tip-to-sample distance and also independent of the bias voltage in narrow energy region $[E_F, E_{F+eV}]$ [52].

Simulated STM images of pristine and defective monolayer of stanene at -1 V are displayed in Figures 4. STM image of pristine stanene shows white (brightest), light brown and

1
2
3 dark brown (darkest) regions, respectively. The distance-height profile clearly indicates the three
4
5 different height patterns corresponding to three different regions in STM image, e.g., positions
6
7 marked as 2, 3 and 7 are recognized as upper atoms, characteristic holes and lower atoms,
8
9 respectively, of stanene in the honeycomb lattice.
10
11

12
13
14 The simulated STM image of the defective stanene also shows three different types of
15
16 regions. The bright (white) region has been seen only for a few atoms which are at the defective
17
18 site and can easily be attributed to the strong out-of-plan reconstruction of atoms. The structure
19
20 of SW-defects can be recognized using the distance-height scan [37]. The distance-height profile
21
22 shows various atoms at different heights, e.g., positions marked as 1, 3 and 5 are recognized as
23
24 one of the upper atom of heptagon, characteristic hole of heptagon and one of the upper atom of
25
26 pentagon, respectively of the defective stanene.
27
28
29

30
31 STM images of all group-V monolayers show similar features [Figure 5]. As a
32
33 representative case, the detailed topography of pristine- and defective-antimonene monolayer is
34
35 shown in Figure 5. Although it is difficult to distinguish between the characteristic hole region
36
37 (marked as 5) and the lower buckled atom (marked as 3) in STM image of pristine antimonene at
38
39 the bias of -1V, but the distance-height profile clearly shows a different height pattern for these
40
41 regions [Figure 5].
42
43
44

45
46 SW-antimonene has distinctly different STM image than those seen for SW-stanene in
47
48 terms of the cross-sectional view and the distance-height profile [Figures 4 and 5]. It is
49
50 important to mention that, for the purpose of comparison, all STM images were obtained using
51
52 the same order of isosurface value (i.e. $10^{-4} e/\text{\AA}^3$) of local density of states, therefore, the colour
53
54 contrast in various images may be different but the general features do not change. We may
55
56
57
58
59
60

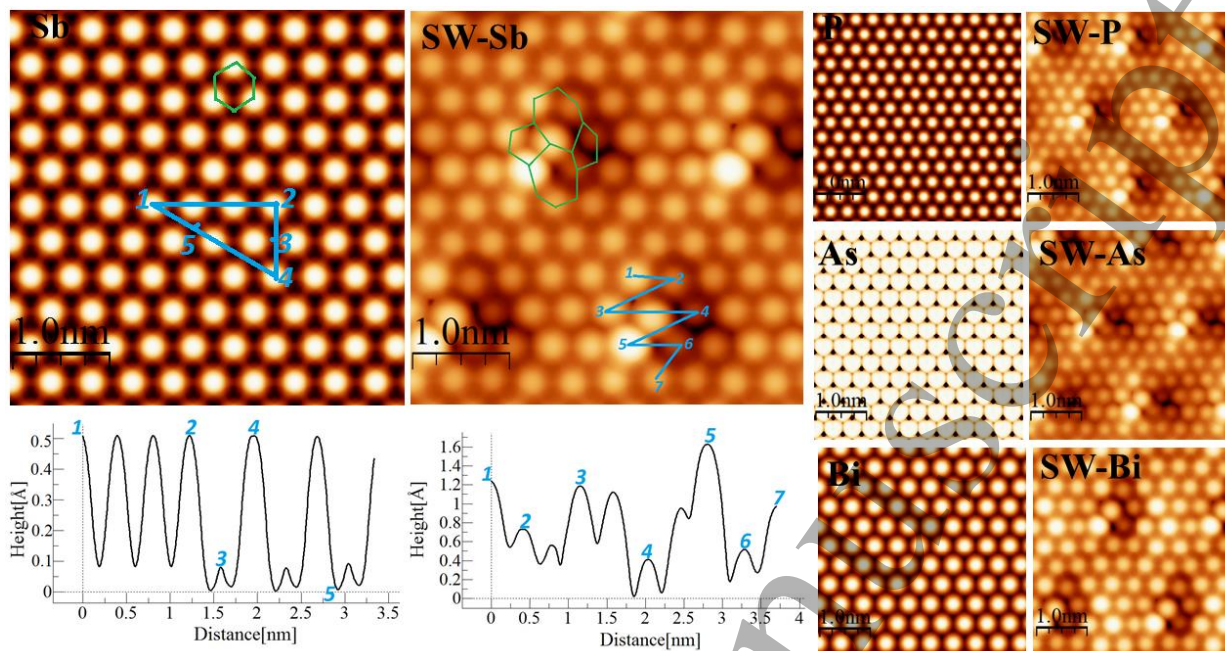


Figure 5: Simulated STM-images of pristine and SW-Sb with distance-height profile along the line shown at bias voltage -1 V. STM images of other group-V monolayer are also given. Images are obtained at constant height corresponding to current imaging mode with isovalue varies as $(0 \text{ to } 0.5) \times 10^{-4} e/\text{\AA}^3$ and $(0 \text{ to } 1.6) \times 10^{-4} e/\text{\AA}^3$ in Sb and SW-Sb, $(0 \text{ to } 0.3) \times 10^{-4} e/\text{\AA}^3$ and $(0 \text{ to } 1.3) \times 10^{-4} e/\text{\AA}^3$ in P and SW-P, $(0 \text{ to } 3.4) \times 10^{-4} e/\text{\AA}^3$ and $(0 \text{ to } 1.6) \times 10^{-4} e/\text{\AA}^3$ in As and SW-As, $(0 \text{ to } 0.6) \times 10^{-4} e/\text{\AA}^3$ and $(0 \text{ to } 1.4) \times 10^{-4} e/\text{\AA}^3$ in Bi and SW-Bi, respectively. Brightest (darkest) regions in STM images are related with the highest (lowest) peak values in corresponding distance-height curve. The order of isovalue is kept same for all the images for the purpose of comparison.

therefore conclude that, considering the topography of defected monolayer, STM may be useful to distinguish between SW-defects in the monolayers of group-IV and group-V elements.

4. Summary

In summary, first principles calculations based on density functional theory are performed to investigate the Stone-Wales defects in monolayers of stanene and group-V elements. Stanene prefers the low-buckled honeycomb structures while group-V monolayers prefer the high-buckled structures in their equilibrium configurations. The SOC effect induces

1
2
3 ~70 meV band gap in stanene by opening up Dirac-cone whereas the band gaps of arsenene,
4 antimonene and bismuthene are reduced by lifting the degeneracy at the valance band maximum.
5
6 At very low bias, both pristine and defective stanene show Ohmic current-voltage characteristics,
7
8 while group-V monolayers act as semiconductors with negligible tunnelling currents. For the
9
10 defective group-V monolayers, an increase in the applied bias increases the tunnelling current
11
12 due to available defect-induced conduction channels in the lattice. Since SW defect-sites can be
13
14 identified using simulated STM images by scanning the distance-height profile, the present study
15
16 may guide experimentalist in characterizing defective monolayers by STM experiments.
17
18
19
20
21
22

23 **Acknowledgements**

24
25
26 PJ is grateful to UGC New Delhi for providing financial assistance in the form of UGC-BSR
27
28 junior research fellowship. Availability of CV Raman (DST, New Delhi, Govt. of India funded)
29
30 cluster at Himachal Pradesh University, Shimla and RAMA High Performance Computing
31
32 cluster at Michigan Technological University Houghton, USA for the present study are
33
34 gratefully acknowledged .
35
36
37
38

39 **References:**

- 40
41
42
43 1. Bhimanapati G R, Lin Z, Meunier V, Jung Y, Cha J, Das S, Xiao D, Son Y, Strano M S,
44
45 Cooper V R, O Liang L, Louie S G, Ringe E, Zhou W, Kim S S, Naik R R, Sumpter B G,
46
47 Terrones H, Xia F, Wang Y, Zhu J, Akinwande D, Alem N, Schuller J A, Schaak R E,
48
49 Terrones M and Robinson J A 2015 *ACS Nano* **9** 11509-11539
50
51
52 2. Zhao A J, Liu H, Yu Z, Quhe R, Zhou S, Wang Y, Liu C , Zhong H, Han N, Lu J, Yao Y
53
54 and Wu K 2012 *Prog. Mater. Sci.* **83** 24-151
55
56
57
58
59
60

- 1
 - 2
 - 3
 - 4
 - 5
 - 6
 - 7
 - 8
 - 9
 - 10
 - 11
 - 12
 - 13
 - 14
 - 15
 - 16
 - 17
 - 18
 - 19
 - 20
 - 21
 - 22
 - 23
 - 24
 - 25
 - 26
 - 27
 - 28
 - 29
 - 30
 - 31
 - 32
 - 33
 - 34
 - 35
 - 36
 - 37
 - 38
 - 39
 - 40
 - 41
 - 42
 - 43
 - 44
 - 45
 - 46
 - 47
 - 48
 - 49
 - 50
 - 51
 - 52
 - 53
 - 54
 - 55
 - 56
 - 57
 - 58
 - 59
 - 60
3. Butler S Z, Hollen S M, Cao L, Cui Y, Gupta J A, Gutiérrez H R and Halperin E 2013 *ACS Nano*. **7** 2898-2926
4. Zhu F F, Chen W, Xu Y, Gao C, Guan D, Liu C, Qian D, Zhang S C and Jia J F 2015 *Nat. Mater.* **14** 1020
5. Saxena S, Chaudhary R P and Shukla S 2016 *Sci. Rep.* **6** 31073
6. Wang D, Chen L, Wang X, Cui G and Zhang P 2015 *Phys. Chem. Chem. Phys.* **17** 26979
7. Wu L, Lu P, Bi J, Yang C, Song Y, Guan P and Wang S 2016 *Nano. Res. Lett.* **11** 525
8. Tang P, Chen P, Cao W, Huang H, Cahangirov S, Xian L, Xu Y, Zhang S C, Duan W, and Rubio A 2014 *Phys. Rev. B.* **90** 121408
9. Zhu Z and Tománek D 2014 *Phys. Rev. Lett.* **112** 176802
10. Kaur S, Kumar A, Srivastava S and K. Tankeshwar 2016 *Phys. Chem. Chem. Phys.* **18** 18312-18322
11. Liu H, Neal A T, Zhu Z, Luo Z, Xu X, Tománek D and Ye P D 2014 *ACS Nano* **8** 4033-4041
12. Lee J, Tian W C, Wang W L and Yao D X 2015 *Sci. Rep.* **5** 11512
13. Zhang S, Yan Z, Li Y, Chen Z and Zeng H 2015 *Angew Commu.* **54** 3112
14. Kou L, Ma Y, Tan X, Frauenheim T, Du A and Smith S 2015 *J. Phys. Chem. C.* **119** 6918-6922
15. Freitas R R Q, Rivelino R, Mota F and Castilho C M C 2015 *J. Phys. Chem. C* **119** 23599-23606
16. Akturk E, Akturk O U and Ciraci S 2016 *Phys. Rev. B* **94** 014115
17. Mardanya S, Thakur V K, Bhowmick S and Agarwal A 2016 *Phys. Rev. B* **94** 035423
18. Kecik D, Durgun E and Ciraci S 2016 *Phys. Rev. B* **94** 205409

19. Ji J, Song X, Liu J, Yan Z, Huo C, Zhang S, Su M, Liao L, Wang W, Ni Z, Hao Y, Zeng Y
2016 *Nat. Commun.* **7** 13352
20. Wang G, Pandey R and Karna S P 2015 *ACS Appl. Mater. Inter* **7** 11490-11496
21. Zhao M, Zhang X and Li L 2015 *Sci. Rep.* **5** 16108
22. Chen H H, Su S H, Chang S L, Cheng B Y, Chen S W, Chen H Y, Lin M F and Huang J C
A 2015 *Sci. Rep.* **5** 11623
23. Stone A J and Wales D J 1986 *Chem. Phys. Lett.* **128** 501-503
24. Hashimoto A, Suenaga K, Gloter A, Urita K and Iijima S 2004 *Nature* **430** 870
25. Bae S, Kim H, Lee Y, Xu X, Park J S, Zheng Y, Balakrishnan J, Lei T, Kim H R, Song Y,
Kim Y J, Kim K S, Ozyilmaz B, Ahn J H, Hong B H and Iijima S 2010 *Nat. Nanotech.* **5**
574
26. Ma J, Alfè D, Michaelides A and Wang E 2009 *Phys. Rev. B* **80** 033407
27. Podlivaev A I and Openov L A 2009 *Phys. Lett. A* **379** 1757-1761
28. Liu H, Feng H, Du Y, Chen J, Wu K and Zhao J 2016 *2D Mater.* **3** 025034
29. Padilha J E and Pontes R B 2016 *Solid. Stat. Commun.* **225** 38-43
30. Hu W and Yang J, 2015 *J. Phys. Chem. C* **119** 20474-20480
31. Liu X, Liu L, Yang L, Wu X and Chu P K 2016 *J. Phys. Chem. C* **120** 24917-24924
32. Hua Y, Wub Y and Zhang S 2016 *Physica B* **503** 126-129
33. Sahin H, Sivek J, Li S, Partoens B and Peeters F M 2013 *Phys. Rev. B* **88** 045434
34. Nardelli M B, Yakobson B I and Bernholc J 1998 *Phys. Rev. B* **57** R4277
35. Yang F O, Huang B, Li Z, Xiao J, Wang H and Xu H 2008 *J. Phys. Chem. C* **112** 12003-
12007
36. Fuhr J D and Saul A 2004 *Phys. Rev. Lett.* **92** 026802

- 1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
37. Gao J, Zhang G and Zhang Y W 2016 *Sci. Rep.* **6** 29107
38. Nigam S, Gupta S, Banyai D, Pandey R and Majumder C 2015 *Phys. Chem. Chem. Phys.* **17** 6705
39. Tersoff J and Hamann D R 1983 *Phys. Rev. Lett.* **50** 1998-2001.
40. Tersoff J and Hamann D R 1985 *Phys. Rev. B* **31** 805-813
41. Bardeen J 1960 *Phys Rev Lett* **6** 57.
42. Lang N D 1986 *Phys Rev. B* **34** 5947
43. Jamdagni P, Kumar A, Sharma M, Thakur A and Ahluwalia P K 2017 *Physica E* **85** 65-73
44. Ahmed T, Kilina S, Das T, Haraldsen J T, Rehr J J and Balatsky A V 2012 *Nano. Lett.* **12** 927-931
45. Kumar A, Banyai D, Ahluwalia P K, Pandey R and Karna S P 2014 *Phys. Chem. Chem. Phys.* **16** 20157
46. Kresse G and Furthmuller J 1996 *Phys. Rev. B* **54** 11169
47. Grimme S 2006 *J. Comput. Chem.* **27** 1787
48. Jamdagni P, Kumar A, Thakur A, Pandey R and Ahluwalia P K 2015 *Mater. Res. Express.* **2** 016301
49. He H, Pandey R, Pati R and Karna S P (2006) *Phys. Rev. B* **73** 195311
50. Pati R, Senapati L, Ajayan P. M. and Nayak S K (2003) *Phys. Rev. B* **68** 100407
51. Sattler K, *Fundamentals of Picoscience* (Taylor and Francis, UK, 2010) p. 415
52. Selloni A, Carnevali P, Tosatti E and Chen C D (1985) *Phys. Rev. B* **31** 2602